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CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N 2 are mixed with 12.0 mol of H 2 according to the following equation: N 2(g) 3H 2(g) 2NH 3(g) N

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Chemistry Chapter 9 Review. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by: sharknerdo. Terms in this set (10) Describe the relationship between Avogadro's number and a mole of a substance. A mole is defined as the amount of a substance in 6.022 x 10^23 (Avogadro's number) particles.

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6 Chemical Bonding  
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