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egativity,

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the Elements AP  
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**Properties of**

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protons in the



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nucleus do not change during normal chemical reactions. Only the outer electrons move.

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know that  
Knowledge is  
actually our  
answer to a more  
rewarding down  
the road, along  
with learning  
doesn't just  
quit once the  
college bell  
rings. That  
becoming  
mentioned, most

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Vocabulary.

Valence  
electron.

Electron dot  
structure. Octet  
rule. Halide  
ion. the

electrons in the  
highest occupied  
energy level of



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an eleme...

diagrams that  
show valence

electrons in the  
atoms of an

elemen... in  
forming

compounds, atoms  
tend to achieve

the electron  
confi...

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equation is

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balanced when the same number of each element is represented on the reactant and product sides. Equations must be balanced to accurately reflect the law of conservation of matter. 3.

Answer Key

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ANSWER Answer  
the following  
questions in the  
space provided.

1. Assign the oxidation number to the specified element in each of the following examples:
- a. S in  $\text{H}_2\text{SO}_3$
  - b. S in  $\text{MgSO}_4$
  - c. S in  $\text{K}_2\text{S}$
  - d. Cu in  $\text{Cu}_2\text{S}$
  - e.

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Cr in  $\text{Na}_2\text{CrO}_4$

5 f. N in  $\text{HNO}_3$

4 g. C in  $(\text{HCO})$

3) 3 h. N in  $(\text{NH})$

4) 2.  $\text{SCl}_2$  a.

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Answer: (b)

Lewis Base.

Explanation:

R-NH<sub>2</sub> (Amines)

behaves as a

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Lewis base

because it is  
capable of

donating a lone  
pair of

electron. We

hope the given

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As shown in  
Figure 7.3.1,  
applying a small  
amount of heat  
to a pile of  
orange ammonium

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dichromate powder results in a vigorous reaction known as the ammonium dichromate volcano. Heat, light, and gas are produced as a large pile of fluffy green chromium(III) oxide forms. We can describe

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this reaction  
with a chemical  
equation An  
expression that  
gives the  
identities and  
quantities of  
the ...

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Principles of

Organic

Chemistry Below

1. What is vital force theory? Who discarded this theory? 2. Which organic compound was first of all prepared in the laboratory?

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is the

kinetically

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product in the following

reaction? A) A

B) B C) C D) D

Ans: A 17. Rank the following carbocations in decreasing order of stability.

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