

# Download Ebook Chapter 22 Hydrocarbon Compounds

## Chapter 22 Hydrocarbon Compounds

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This hydrocarbon is called "2,3-pentadiene" because there are 2 (di) double bonds (ene) after 2 and 3 (2,3) of a 5 carbon line (penta). It is named the same way for triple bonds, but with an yne instead of ene. Next are isomers. Isomers are compounds with the same molecular formula but different molecular structures.

Chapter 22: Hydrocarbon Compounds - Chemistry by Anna molecules of hydrocarbons, such as alkanes, are nonpolar molecules. Hydrocarbons. the simplest organic compounds contain only carbon and hydrogen. Alkane. a hydrocarbon in which there are only single covalent bonds. Straight-chain alkanes. contains any number of carbon atoms, one after the other, in a chain.

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Section 1: Hydrocarbons Hydrocarbons: Organic compound that contains only carbon and hydrogen. Alkane:

Hydrocarbon containing only single covalent bonds. Straight-chain: Saturated hydrocarbon that contains any number of carbon atoms arranged one after the other in a chain.

Homologous Series: Group of compounds in which there is a constant increment of change in molecular structure from one ...

chapter 22: hydrocarbon compounds

Made up of about 80% methane, 10% ethane, 4% propane, and 2% butane. The remaining 4% consists of nitrogen, helium, and hydrocarbons of higher molar mass. Propane and butane are separated from the other gases in natural gas by liquefaction. Complete combustion of a hydrocarbon produces a blue flame.

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Liv20cederstrom. Terms in this set (23) Alkanes. a

hydrocarbon containing only single covalent bonds. Alkenes.

Hydrocarbons with one or more carbon-carbon double bonds. alkyl group. a hydrocarbon substituent.

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an organic compound that contains only carbon and  
hydrogen. alkane. a hydrocarbon containing only single  
covalent bonds; saturated hydrocarbons. straight-chain  
alkane. a saturated hydrocarbon that contains any number of  
carbon atoms arranged one after another in a chain.  
homologus series.

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1. Find the root word (ending in -ane) in the hydrocarbon  
name. Then write the longest carbon chain to create the  
parent structure. 2. Number the carbons on this parent  
carbon chain. 3. Identify the substituent groups. Attach the  
substituents to the numbered parent chain at the proper  
positions. 4. Add hydrogens as needed.

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Chapter 22 "Hydrocarbon Compounds". Use these activities  
to learn the vocabulary and major concepts presented in the  
chapter. molecules in which the atoms are joined in the same  
order, but the arrangements of the atoms in space are  
different.

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SECTION 22.1 HYDROCARBONS(pages 693–701)

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Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Wendy\_Alexander3. Terms in this set (20) Organic Compound. Applied to all carbon containing compounds with the primary exceptions of carbon oxides, carbides, and carbonates, which are considered inorganic. Organic Chemistry is ...

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This unit looks at aliphatic and aromatic hydrocarbons, including alkyl substituted hydrocarbons and structural and geometric isomers. There is also a brief introduction to functional groups. Chapter 22 - Hydrocarbon Compounds

Chapters 22 & 23 - Organic Chemistry - Mrs. Gingras ...  
Hydrocarbon compounds Section 22.1: Hydrocarbons What is the relationship between carbon and organic chemistry? 200 years ago scientists thought that only living organisms could make carbon compounds known in living cells Organic means related to or made by living things In 1828, A German Chemist, Friedrich Wohler

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